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Chapter 9

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continued 18. When a strong acid is
titrated with a weak base, the pH of

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the solution at the equivalence point is
than 7. 19. A is a highly purified solid
used to check the concentration of a
standard solution. 20. A 1 M solution
of NaOH will have a pH that is

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MARSHALL'S CLASSROOM
Chemistry (4th Edition) Burdge, Julia
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This chapter and the chapters that follow provide you with the tools necessary to answer these questions and many others like them.

9.1 A Typical Problem
9.2 Relating Moles to Number of Particles
9.3 Molar Mass and Chemical Compounds
9.4 Relationships Between Masses of Elements and Compounds
9.5 Determination of Empirical and

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