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The plot of  $V = V(P, T)$  for a gas is

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shown in Fig. 1.1. Determine the expressions of the two second derivatives of the volume of this plot. (note: the principle curvatures of the surface are proportional to these second derivatives).

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Solution: A volume of 1 L equals 0.001 m<sup>3</sup>, see Table A.1. From Figure 2.7 the density is in the range of 10 so we get  $m = 10 \cdot 0.001 \text{ m}^3 = 10 \text{ kg}$ . A more accurate value from Table A.4 is 13 580. For the air we see in Figure 2.7 that density is about 1 so we get  $m = 1 \cdot 0.001 \text{ m}^3 = 0.001 \text{ kg}$ . A more accurate value from Table A.5 is 1.17 at 100 kPa, 25°C.

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