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Chemistry Live! – Worked Solutions 2 Chapter 9 - The Mole Concept 9.1 Since a sulfur atom (Ar = 32) weighs twice as much as an oxygen atom (Ar = 16), 32 g of sulfur atoms and 16 g of oxygen atoms both contain the same number of atoms. Answer: 32 g 9.2 (a) 1 mole of H atoms = 1 g 1 mole of H2 molecules =  $2 \times 1 = 2$  g. (b) 1 mole of N atoms = 14 g

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