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4NH (g) 5O (g) 4NO(g) 6H O(g) kJ mol 2 2+ + = – H O 910 – 1 Describe the effect, if any., of each of the following changes on the equilibrium concentration of nitrogen monoxide in a particular equilibrium mixture, giving a reason in each case. (i) Increasing the pressure, at constant temperature [2]

CHEMISTRY STANDARD LEVEL PAPER 2

– 2 – Section A Answer all questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid, HOOC – COOH • xH₂O, by titrating a known mass of the acid with a 0.100 mol dm⁻³ solution of NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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Ethane, C₂H₆, and disilane, Si₂H₆, are both hydrides of group 4 elements with similar structures but with different chemical properties. (a) (i) Deduce the Lewis (electron dot) structure for Si₂H₆ showing all valence electrons. [1] (ii) State and explain the H – Si – H bond angle in Si₂H₆. [2]

CHEMISTRY Candidate session number STANDARD LEVEL PAPER 2

[2] (b) Marit arranged the values she found in Table 12 into an energy cycle. C₂H₄(g) + H₂(g) C₂H₆(g) 3 O 2 2CO₂(g) + 3H₂O – 1 4 1 1 k J m o l – 1 – 2 8 6 k J m o l – 1 – 1 5 6 0 k J m o l – 1 (hydrogenation) (l) Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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