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STUDY. PLAY. single-displacement.
the replacement of hydrogen from
water by sodium is an example of a
_____ reaction. activation energy. in
order for a chemical reaction to take

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place, the particles involved must collide with a sufficient amount of

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1) calculate moles of each element (if given mass) 2) divide each mole

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quantity by the smallest mole value from step 1. 3) use whole number values from step 2 as the subscripts to the original equation to write the empirical formula. Percent Composition. -the relative amounts of each element in a compound.

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Ionic Bond. Nonpolar Covalent Bond.
Polyatomic Ions. unit of matter that
results when 2 or more atoms are
joined by.... A bond resulting from the
electrostatic attraction of a cation.... A

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bond characterized by an even
distribution of charge; electr....

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Multiple Choice 24) What is the

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amount of heat required to raise the temperature of 200.0 g of aluminum by 10°C? cal (specific heat of aluminum = 0.21 J/g°C)

24) A) 4200 cal B) 420 cal C) 42 000 cal D) 420 000 cal E) none of the above

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2nd Semester Chemistry-575 Final
Exam Review Multiple Choice Identify
the choice that best completes the
statement or answers the question.

_____ 1. How many moles of tungsten
atoms are in 4.8×10^{25} atoms of
tungsten? a. 8.0×10^2 moles c. $1.3 \times$

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10²¹ moles b. 8.0×10^1 moles d. 1.3×10^2 moles _____ 2.

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Chemistry had a bad reputation because of alchemy, which was a pseudoscience that claimed to convert cheap metals into gold. Alchemy led to the discovery of many elements. 1500 – George Bauer discovered

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metallurgy, extracting pure metal out
of a mix 1600s – Robert Boyle studied
the behavior of gases and is
responsible for the theory of elements
...

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