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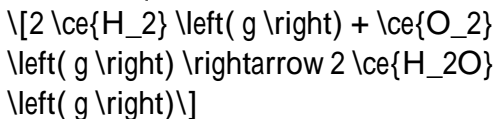
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and Equations - Lesson 05 | Balancing
Chemical Equations Part 2

Chemical Equations Reactions Section 2

1. They could draw a picture of the chemical
reaction. 2. They could write a word
equation for the chemical reaction: "Two
molecules of hydrogen gas react with one
molecule of oxygen gas to produce two
molecules of water vapor." 3. They could
write the equation in chemical shorthand.



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Answers

8.2: Chemical Equations - Chemistry

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Chemical Equations and Reactions

SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1.

Match the equation type on the left to its representation on the right. c synthesis (a)

$AX + BY \rightarrow AY + BX$ d decomposition (b) A

$BX + AX \rightarrow B + X$ b single-displacement (c) A B

$AX + BX \rightarrow AX + BX$ a double-displacement (d) $AX + A \rightarrow X$

8 Chemical Equations and Reactions

Here is the balanced symbol equation: $2Cu + O_2 \rightarrow 2CuO$. You can see that we now

have two copper atoms and two oxygen atoms on each side. This matches what

happens in the reaction:

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Chemical equations - Types of reaction -
KS3 Chemistry ...

Interactive Textbook 23 Chemical Reactions
SECTION 2 Name Class Date Chemical
Formulas and Equations continued HOW
TO BALANCE A CHEMICAL
EQUATION To balance an equation, you
must use coefficients. A coefficient is a
number that is placed in front of a chemi-cal
formula. For example, 2CO represents two
carbon monoxide molecules.

2 SECTION 2 Chemical Formulas and
Equations

Interactive Textbook 250 Chemical
Reactions SECTION 2 Name Class Date
Chemical Formulas and Equations
continued CHECKING SYMBOLS When
you write a chemical formula, it is important
that you check to make sure that it is correct.

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Chemical Equations

If you use the wrong formula or symbol in an equation, the equation will not describe the correct reaction. Even a small mis-

14 SECTION 2 Chemical Formulas and Equations

This is second lecture on a CBSE Class 10th topic - Chemical Reactions and Equations. This chapter is divided into 3 lectures, so make sure you watch the oth...

Chemistry: Chemical Reactions and Equations (Part 2 ...

the general equation for a single displacement reaction is displacemnt of the metal in a compiund by another metal, displacemnet of the halogen in a compound by another halogen, dispolacemnt of the hydrogen in water by a meatl species, the displacemnt of a hydrogen in acid by a metal

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chapter 8 chemical equations and reactions
section 2 ...

Section 2: Chemical equations. STUDY.
Flashcards. Learn. Write. Spell. Test. PLAY.
Match. Gravity. Created by.

LaurenMiley2000. Terms in this set (13)
chemical equation. are shorter, easier, way to
show chemical reactions, using symbols
instead of work. What's needed to write an
equation - chemical formulas, reactants, and
products. chemical ...

Section 2: Chemical equations Flashcards |
Quizlet

4.1: Chemical Reactions and Chemical
Equations Last updated; Save as PDF Page
ID 24181; No headers $[2\text{H}_2 + \text{O}_2$
 $\rightarrow 2\text{H}_2\text{O}]$ Chemical formulas

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Chemical Equations

and other symbols are used to indicate the starting materials, or reactants, which by convention are written on the left side of the equation, and the final compounds, or products, which are written on the right.

4.1: Chemical Reactions and Chemical Equations - Chemistry ...

Types of reaction Atoms are rearranged during chemical reactions, and are not lost or gained. Chemical reactions can be represented using equations. Catalysts speed up reactions without being used...

Chemical reactions - Types of reaction - KS3 Chemistry ...

Ch. 8 Chemical Equations And Reactions 1. Chemical Equations and Reactions CHAPTER 8 2. What is a chemical equation? 3. List and describe some

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indications of an occurrence of a chemical reaction. 4. List and describe some indications of an occurrence of a chemical reaction.

Ch. 8 Chemical Equations And Reactions

CHEMICAL EQUATIONS AND

REACTIONS 261 SECTION 1 O

BJECTIVES List three observations that suggest that a chemical reaction has taken place. List three requirements for a correctly written chemical equation. Write a word equation and a formula equation for a given chemical reaction. Balance a formula equation by inspection. Describing Chemical ...

CHAPTER 8 Chemical Equations and

Reactions

Chemical Equations and Reactions

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SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right. a b synthesis decomposition single-displacement double-displacement (a) $AX + BY \rightarrow AY + BX$ (b) $A + BX \rightarrow AX + B$

Home - Kenilworth Public Schools
Chemical reactions are represented by balanced chemical equations, with chemical formulas symbolizing reactants and products while chemical equations give information in two major areas: first, they tell us what substances are reacting (those being used up) and what substances are products (those being made).

Chemical Reactions and Chemical Equations - Owlcation ...

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Chemical Equations

Section 2 Chemical Formulas and Equations

Key Concept Chemical formulas and chemical equations are used to show how atoms are rearranged to form new substances in a chemical reaction. •

Chemical formulas are a simple way to describe which elements are in a chemical substance. • Chemical equations are a concise way to write how atoms are

Section 2 Chemical Formulas and Equations

The five general types of reaction are combination, decomposition, single-replacement, double-replacement, and combustion. Not all chemical reactions fit uniquely into only one category.

Occasionally, a reaction may fit equally well into two categories. Nevertheless, recognizing a reaction as a particular type is useful.

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11.2 Types of Chemical Reactions - Henry County School ...

It can be balanced by putting a 2 before O_2 , giving the balanced equation: $CH_4 + 2 O_2$

$CO_2 + 2 H_2O$. This equation does not have any coefficients in front of CH_4 and CO_2 , since a coefficient of 1 is dropped.

Matrix method. Generally, any chemical equation involving J different molecules can be written as:

Chemical equation - Wikipedia

- Finish Predicting Reactions- worksheet #1

(* ANSWERS) *Extra Practice:

Combustion Read 3.3 (p. 137 - 141),

ANSWER p. 141 #41 - 50: Wed, Oct 9:

Continue work from yesterday: Predicting

Reactions * Worksheet #2 - Predicting the

Products of Chemical Reactions

SUMMARY- Types of Decomposition

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