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substances taking part in a reaction or forming a compound. What is stoichiometry? Composition deals with the mass relationships of elements in compound; reaction involves the mass relationships between reactants and products in a chemical reaction. What is the difference between reaction ...

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CHAPTER 9 REVIEW Stoichiometry SECTION 3

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of  $N_2$  are mixed with 12.0 mol of  $H_2$  according to the following equation:  $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$  ...

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9-1 Introduction to Stoichiometry Composition  
Stoichiometry - deals with mass relationships of  
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products in a chemical reaction I. Page 4/15

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