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Lecture 4: Chemical  
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**COVALENT**

**BONDING** Chemical  
Bonding - Ionic vs.  
Covalent Bonds

**Concept of Valency -  
Introduction | Atoms  
And Molecules |**

**Don't Memorise** Ionic  
and Covalent Bonds,  
Hydrogen Bonds, van  
der Waals - 4 types of  
Chemical Bonds in  
Biology *Ionic and*

*Covalent Bonds Made*

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*Hydrocarbons |*

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**1: Ions** *The Periodic*



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*Table: Atomic Radius,  
Ionization Energy,  
and Electronegativity*

**Valency | Covalent**

**bond | Chemical**

**bonding (part 3) | 9th**

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**CGBSE | SCERT |**

**G.Sc. Covalent**

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### Answer

When H forms a bond with H O to form the hydronium ion H O , this bond is called a coordinate covalent bond because a. both bonding electrons come from the oxygen atom. b. it forms an especially strong bond. c. the electrons are equally shared. d. the oxygen no longer

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has eight valence electrons.

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Authors: Wilbraham,

ISBN-10:

0132525763,

ISBN-13:

978-0-13252-576-3,

Publisher: Prentice

Hall

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angle of 109.5 degrees that results when a central atom forms four bonds directed toward the center of a regular tetrahedron. the mixing of several atomic orbitals to form the same total number of equivalent hybrid orbitals.

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*Page 15/29*

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**Bonding Crossword  
Puzzle Answers**

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Section 8.1 The

Covalent Bond pages

240–247 Practice

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Draw the Lewis

structure for each

molecule. 1.  $\text{PH}_3$

$\text{H}-\text{H}-\text{H}$   $\text{H}-\text{P}$

respectively, for

single, double, and

triple  $\text{P}=\text{P}$  2.  $\text{H}_2\text{S}$

$\text{H}-\text{H}$   $\text{H}-\text{S}-\text{S} \dots$

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8.2 8.2 8.2 8.4 8.3 8.3

8.3 195 Chapter Quiz

Choose the best answer and write its letter on the line. 1. A bond in which each atom contributes two electrons is a. a double covalent bond. b. an ionic bond. c. a polar covalent bond. d. a coordinate covalent bond. 2. The electron dot structure

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for hydrogen sulfide,  
 $H_2S$  ...

Bonding

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?questionWhen  
sharing of electrons  
occurs the attachment  
between atoms is  
called answercovalent  
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dissociation energy is

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Covalent Bonding Or  
How I Learned to  
Love Sharing (But  
Remember, File  
Sharing is Illegal) 2.

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remember, ionic compounds are solids at room temperatures that have one ion strip the electron (s) from the other elements' electron cloud.

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8-11. covalent nuclei. This results in stronger attractive forces between electrons and nuclei, decreasing the distance between the nuclei. A carbon-carbon single bond has a bond order of 1 and is longer than a carbon-carbon double bond with a bond order of 2.

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Covalent

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Covalent Bonding

Answer Key -

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molecular structure.

## Covalent Bonding

## Answer Key

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section 81 molecular  
compounds answers  
could add your near  
connections listings.

### **Chapter 8 Covalent Bonding Section 81 Molecular Compounds ...**

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Bonding. Matter takes

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many forms in nature:

In this chapter, we are going to learn to

distinguish the type of compound that we

have already studied, the “ionic compound”

(which contains

oppositely-charged

particles: metal

cations and non-metal

anions), from a

different type of

compound – a

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“molecular  
compound”.

### Chapter 8: Covalent Bonding

Chapter 8 - Covalent  
Bonding/Polarity Quiz  
Answer Section

MULTIPLE CHOICE

1. ANS: D PTS: 1

DIF: L3 REF: p. 248 |

p. 249 OBJ: 8.4.1

Describe how  
electronegativity

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values determine the charge distribution in a polar molecule.

MSC: application

2.ANS: B PTS: 1 DIF:

L1 REF: p. 251

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