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11 chap 6 | Thermodynamics 07 || Heat of Reaction | Enthalpy Of Formation | Enthalpy Of Combustion | ~~90 Minutes of Thermo/Enthalpy/Heat Practice~~ Chapter 17 Thermochemistry Worksheet Answers

As liquids absorb heat at their boiling points, the temperature remains constant while they vaporize. true _____ Chapter 17 Thermochemistry 187 05_Chem_GRSW_Ch17.SE/TE 6/11/04 3:49 PM Page 188 Name _____ Date _____ Class _____ CHAPTER 17, Thermochemistry (continued) Use the heating curve for water shown below to answer Questions 5, 6, and 7.

section 17.1 the flow of energy heat and work

Chapter 17 Thermochemistry Packet Answers 2. 0.100 g of H2 and an excess of O2 are compressed into a bomb calorimeter containing 1200 g of water. The temperature before the reaction is 25.00 °C, and after the reaction it goes to 27.16 °C.

Chapter 17 Thermochemistry Packet

CHAPTER17, Thermochemistry (continued) Use the heating curve for water shown below to answer Questions 5,6, and 7. Heating Curve for Water Boiling point ~::>,-----,1 ~ Melting E point ~ <--_----'I Heat supplied 5. Label the melting point and boiling point temperatures on the graph. 6. What happens to the temperature during melting and vaporization?

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Chapter 17 Thermochemistry187 10. Complete the enthalpy diagram for the combustion of natural gas. Use the thermochemical equation in the first paragraph on page 517 as a guide. SECTION 17.3 HEAT IN CHANGES OF STATE (pages 520□526) This section explains heat transfers that occur during melting, freezing, boiling, and condensing.

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505□510)
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