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4. true 5. true 6. 2, 2, 64.10 7. 3, 3,
96.00 8. 2, 2, 88.02 9. 4, 4, 72.08 10.
methanol and oxygen gas 11. carbon
dioxide and water 12. 160.10 g 13.
160.10 g 14. They are equal. 15. A
mole ratio is a ratio between the
numbers of moles

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Section 11.1.1 Objectives Describe
the types of relationships indicated by
a balanced chemical equation. State
the mole ratios from a balanced

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Chemical equation. Review Vocabulary
reactant: the starting substance in a
chemical reaction New Vocabulary
stoichiometry mole ratio Defining
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Answer Key

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Guide In Section 11.3, for example, you learned how to express the stoichiometry of the reaction for the ammonium dichromate volcano in terms of the atoms, ions, or molecules involved and the numbers of moles, grams, and formula units of each (recognizing, for instance, that 1 mol

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of ammonium dichromate produces 4 mol of water). This section describes how to use the stoichiometry of a reaction to answer questions like the following: How much oxygen is needed to ensure complete combustion of a ...

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