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Date Period Purpose To examine the arrangements of electrons in subshells. 1. Use the drawings to help
you complete the table. Write the number of electrons in each subshell in the boxes. 1s 2s 2p 3s 3p 3d 4s
4p 4d 4f Carbon 2e 22e2 2e aluminum potassium Bromine LESSON 24 cLASSWOrk
LBCTCM_01_BFW1st12.indd ...

LESSON 24 Shell Game Name Electron Configurations

The unit used for expressing atomic mass. $1 \text{ amu} = 1.66 \times 10^{-24} \text{ g}$, the mass of one hydrogen atom.

This is 1/12 the mass of a carbon-12 atom. periodic table of the elements A table with elements
organized in order of increasing atomic number, and grouped such that elements with similar properties
are in vertical columns.

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